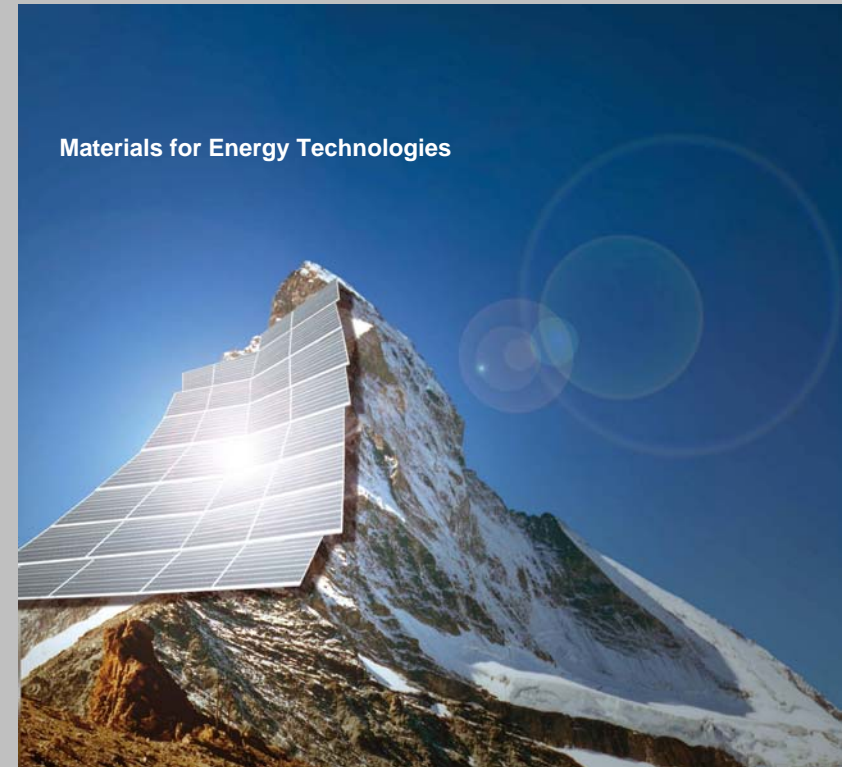


Life Cycle Assessment LCA of Li-Ion batteries for electric vehicles

1. Project goals
2. Presentation of a typical battery
from cradle to gate
3. LCA results



graphics: Empa

M. Gauch, R. Widmer, D. Notter, A. Stamp, H.J. Althaus, P. Wäger

Empa - Swiss Federal Laboratories for Materials Testing and Research
TSL Technology and Society Lab

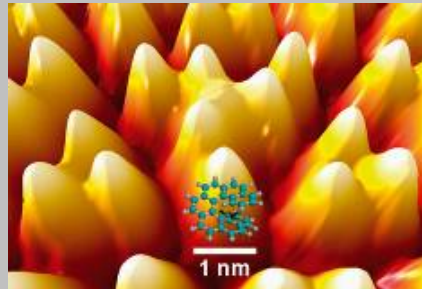
2009



Materials Science & Technology

Empa's Research Programs

Materials for Health & Performance

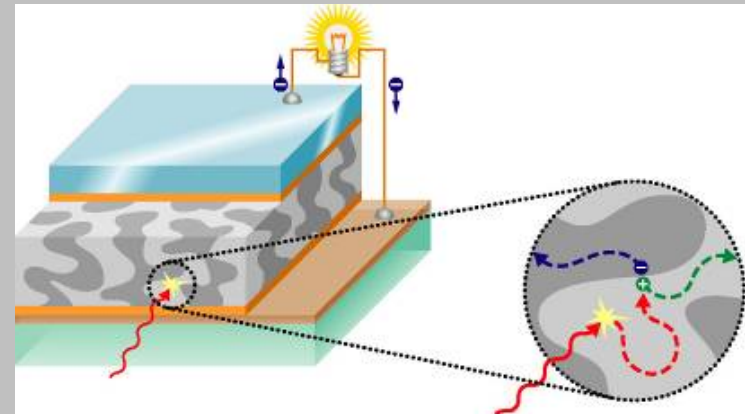


Nanotechnology

Adaptive Material Systems



Technosphere Atmosphere



Materials for Energy Technologies

Technology and Society Lab @ Empa

- Life Cycle Assessments
- Scarce materials resources
- Energy options for transitional countries

Project goals

- Detailed cradle-to-gate Life Cycle Inventory LCI of a modern Li-Ion battery
- Special focus on Lithium mining and refining to battery-grade material
- Integration of the results in ecoinvent database
- Evaluation of the environmental impacts with Life Cycle Assessment LCA tools
- how harmful is the battery / a vehicle / a km in comparison with a standard ICE car?

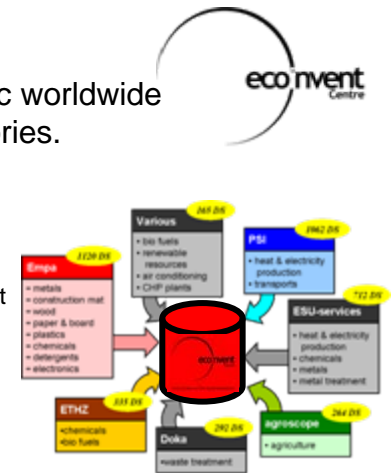


4 In-wheel motors (each 30 kg), > 400 PS / 160 km autonomy
Removal of drivetrain components: 1000 pounds (454kg)
Batteries Lilon 1000 pounds (40kWh),
Prototype for tuningfare SEMA Las Vegas Nov '08

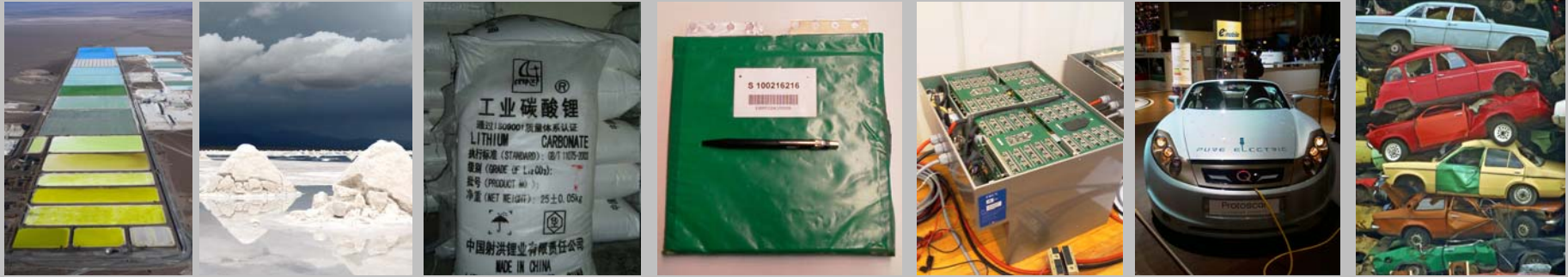
ecoinvent

The most comprehensive public worldwide database for Life Cycle Inventories.

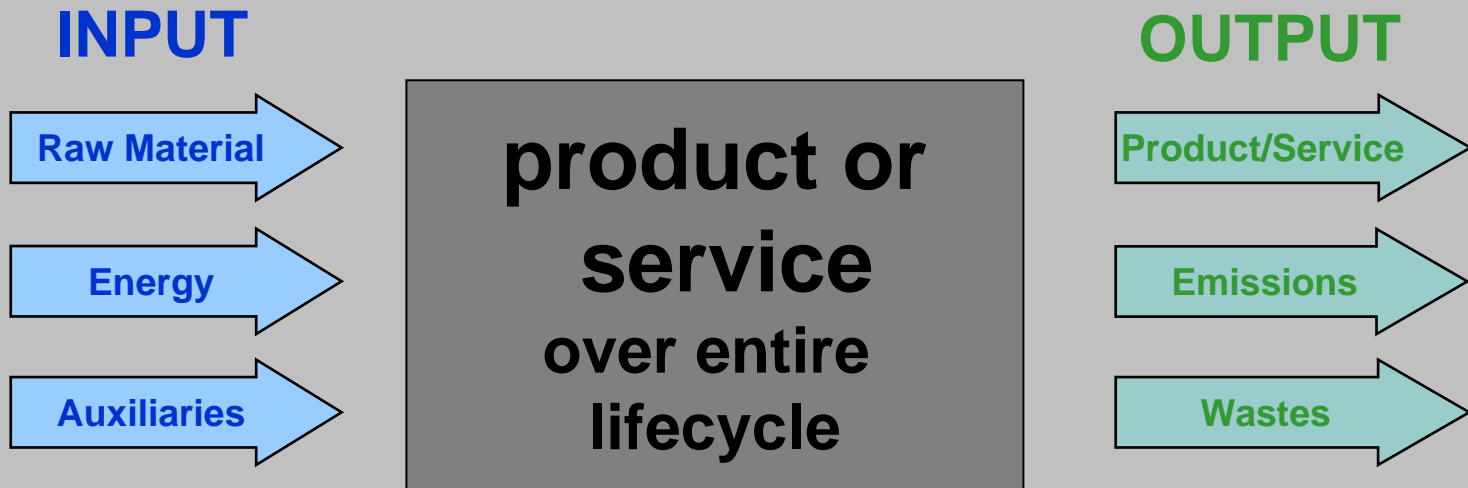
- More than 4000 process on energy supply, resource extraction, material supply, chemicals, metals, agriculture, waste management services, transport services
- Based on industry data, compiled by independent experts
- Consistent, validated and transparent



Life cycle of a battery

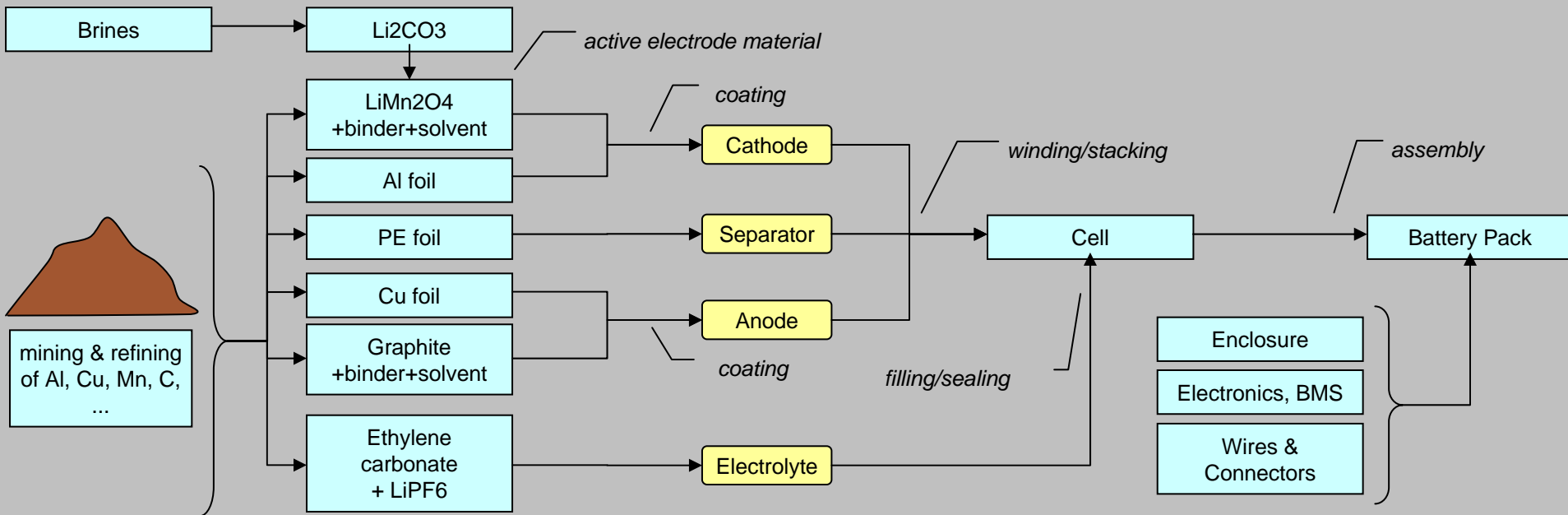


Life cycle assessment: The basic idea



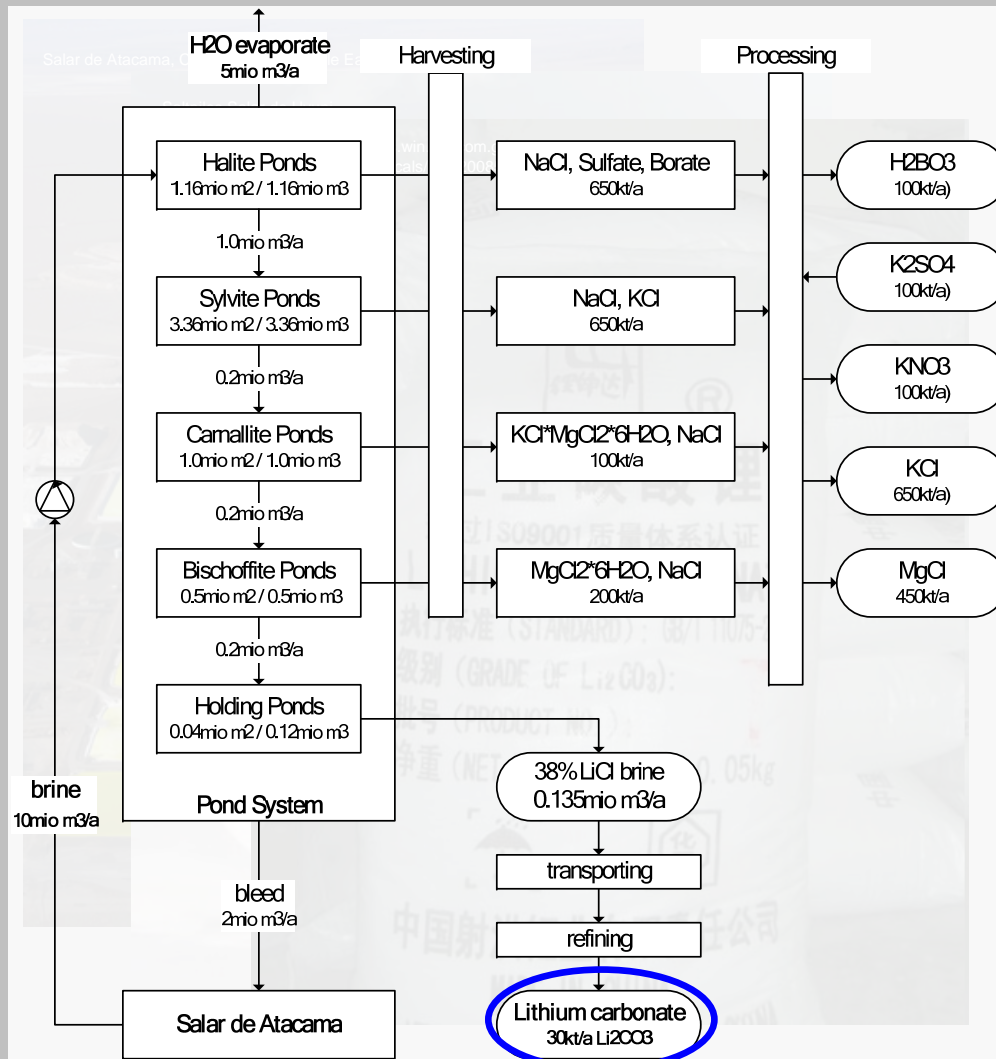
& ecological assessment of flows

A Li-Ion Battery is built (Manganese oxide type)



- Two foils and a thin separator that is permeable to ions are wound or folded into a stack to pack a big surface into a low volume
- This stack is inserted in a pouch or case, filled with Li-salt electrolyte and sealed.
-> The **cell** is ready (assumption for this study: 3.7V; 40Ah; 130 Wh/kg; 5C_{cont}.)
- These cells are combined in series or parallel and completed with a battery management system (BMS) and the necessary wirings. This assembly is fit into an enclosure
-> The **battery pack** is ready
- Assumption for this study: 100Wh/kg, 5C_{cont}, 30kWh, 300kg

Lithium



- Lightest metal (density 0.543 kg/l)
- Highest electrochemical potential
- Relatively abundant (more abundant than Cu, seawater content 0.17 ppm)
- Non-toxic (used as drug)
- Very reactive in metallic form (burns!)

Lithium exploitation

- Most of the lithium for batteries comes from saltlakes in the Andes (Chile, Bolivia) or in China (Tibet)
- It's extracted from salines and sold as lithium carbonate Li_2CO_3
- The highest energy fraction for the production of Li_2CO_3 is solar energy for the evaporation
- Most of the by-products go to fertilizer production

Extraction of lithium carbonate from Atacama (CL), one of the most important worldwide Li_2CO_3 producers
 Compilation of numbers and graphics: Empa

A Li-Ion Battery Cell (Manganese oxide type)



Cell calculation		per kg cell
Total cell pack		[g]
Cathode	Al collector foil	143.7
	Cathode Li-X	240.8
Anode	Cu-collector foil (spec.sheet: 8-15um)	124.8
	Anode graphite	162.3
Separator	Separator film PE	51.4
Packing	PE-Al envelope	70.2
Electrolyte	Ethylencarbonate (w/o LiPF6 1M)	171.2
Electrolyte Salt	LiPF6 1M (152 g/mol, 1mol/l)	19.7
Electrodes	Electrode tabs Al	15.8
Total cell pack		1000.0

Table: measurements Empa

What's in it?

- Only ~1% of a Li-Ion cell is Li (resp. 5% Li_2CO_3) which means 0.08 kg Li for 1kWh
- ~40% of a cell is Al (~23%) and Cu (~13%) (current transducer, electrode carrier)
- ~40% is the active electrode material (cathode LiMn_2O_4 ~24%, anode graphite ~16%)
- ~20% is the electrolyte (lithium hexafluorophosphate LiPF_6 1M in solvent ethylencarbonate)
- The metals Cu, Al, Mn/Co/Ni/Fe are usually recycled
- Graphite, electrolyte and Li usually are not recycled for cost and energy efficiency reasons

How to measure the env. impacts? How 'clean'?

reasonable answer needs:

- A good bookkeeper, who
 - Sums up all the inputs
 - Defines the system limits
- Help from software with a database which contains all the relevant values
- Different environmental impacts can be evaluated (e.g. greenhouse-gas, air/water/soil-emissions, landuse,...)
- Methods which allow to conclude all the impacts to a total environmental impact figure



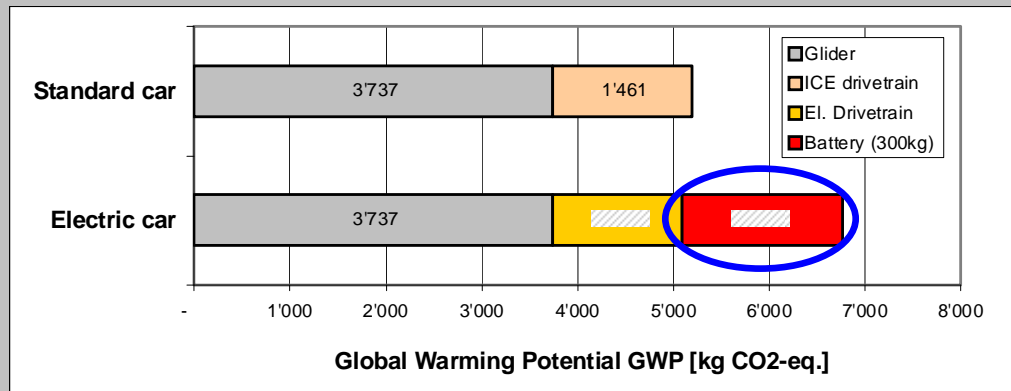
Result of this bookkeeping over the whole lifecycle:
-> **Ecobalance**

LCA result Comparison: ICE / BEV vehicle production

Global warming potential GWP (kg CO₂-eq) of the car production (w/o operation)

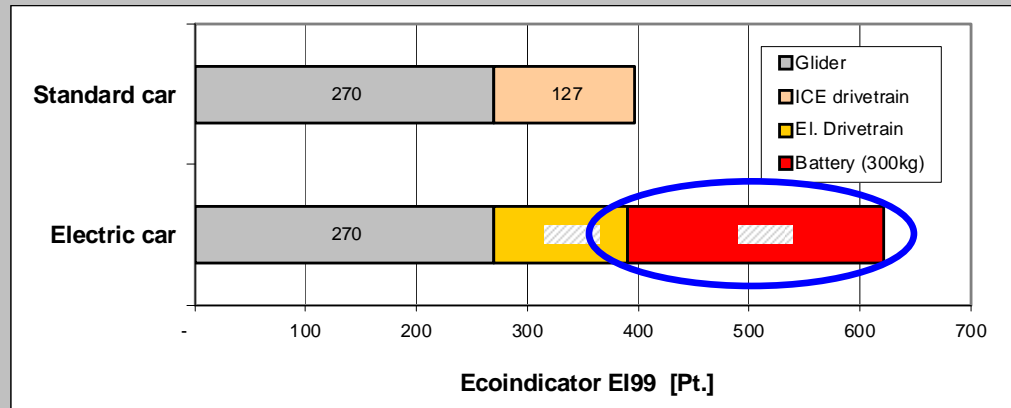
ICE, golf-class

BEV, golf-class,
300kg battery 30kWh
~200km autonomy in NEDC



plus ~30%

Total environmental impact (ecoincicator EI99 points) of the car production (w/o operation)



plus ~57%

- Producing a BEV incl. battery causes significantly more damage than a conventional ICE car
- Producing an electric drivetrain w/o battery causes slightly less damage than an ICE drivetrain

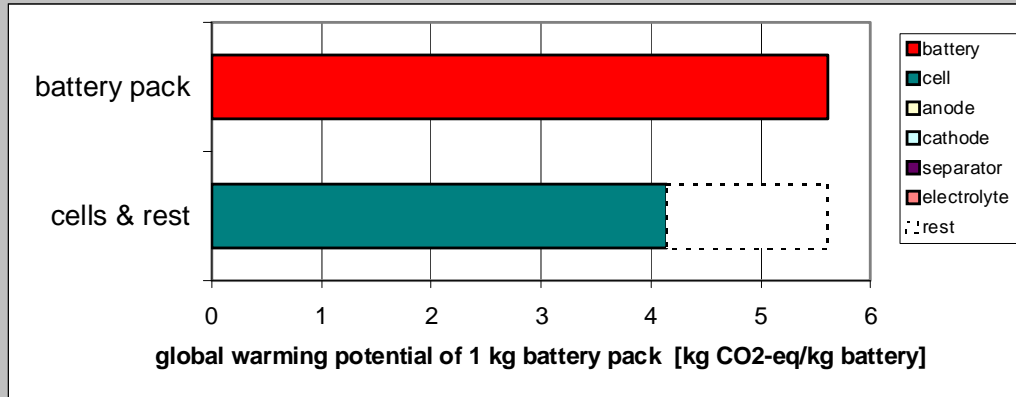
LCA result Li-Ion battery (manganese oxide type)

Global warming potential (kg CO₂-eq) for the production: 1 kg battery

battery pack composition

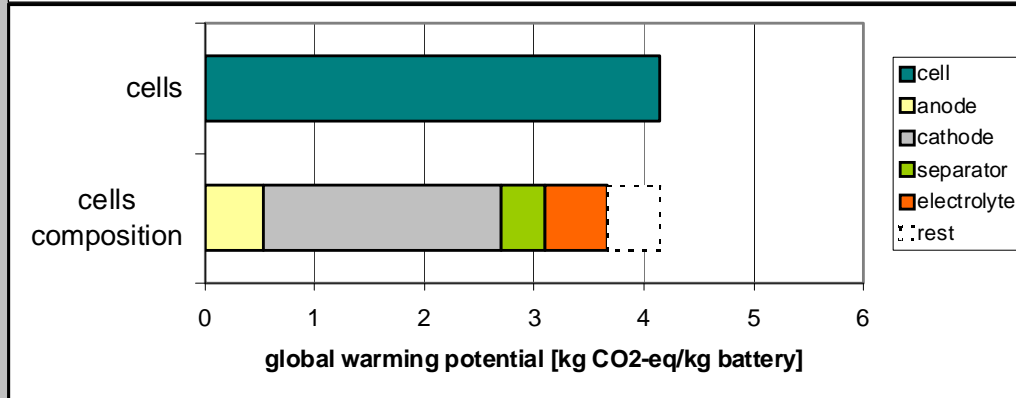
cells and assembly to a battery pack

- 74% from cells
- rest: assembly plant, BMS printed circuit board, wirings, enclosure, production energy, transport



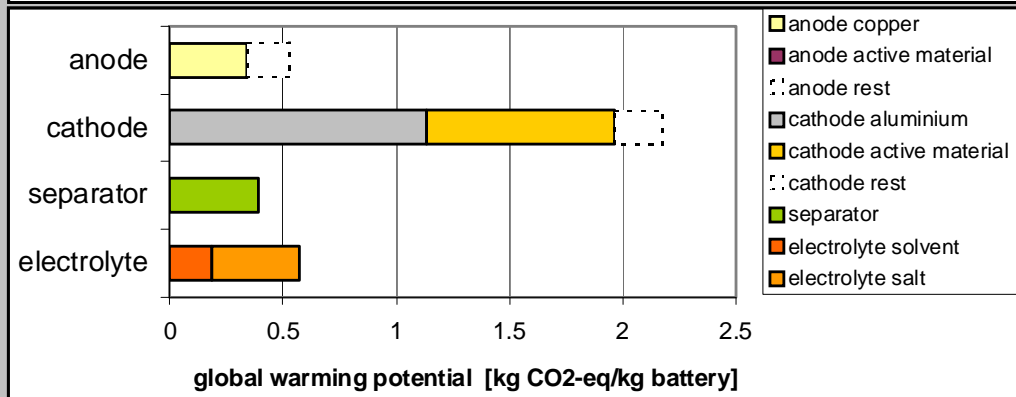
cell composition

- main impact 52% from cathode
- rest: cell plant, pouch (plastic), production energy, transport



cell details

- highest impact from Al and LiMn₂O₄ in cathode
- similar impacts from Cu, separator and electrolyte salt LiPF₆



conclusion:
optimisation of metals use



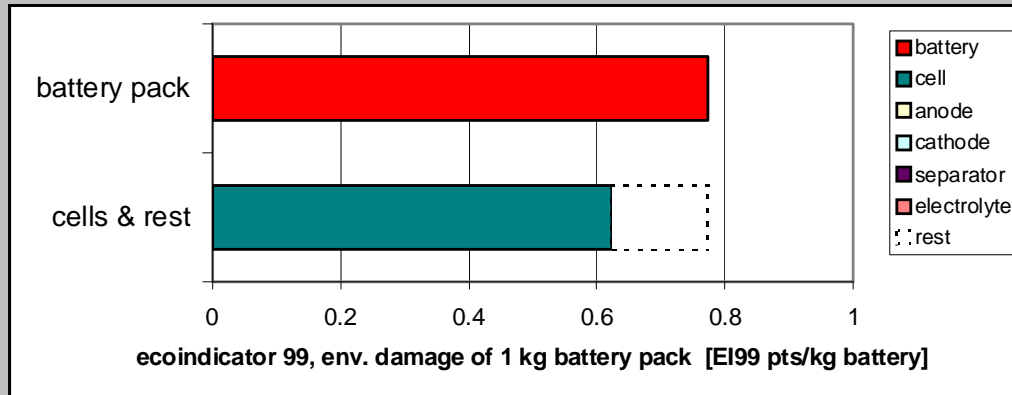
LCA results Li-Ion battery (manganese oxyde type)

Total environmental impact (ecoindicator EI99 pts) for the production: 1 kg battery

battery pack composition

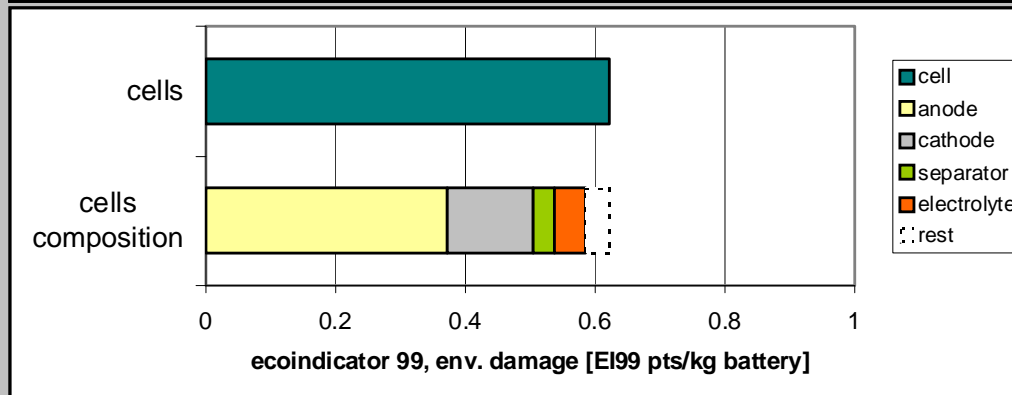
cells and assembly to a battery pack

- 81% from cells
- rest: assembly plant, BMS printed circuit board, wirings, enclosure, production energy, transport



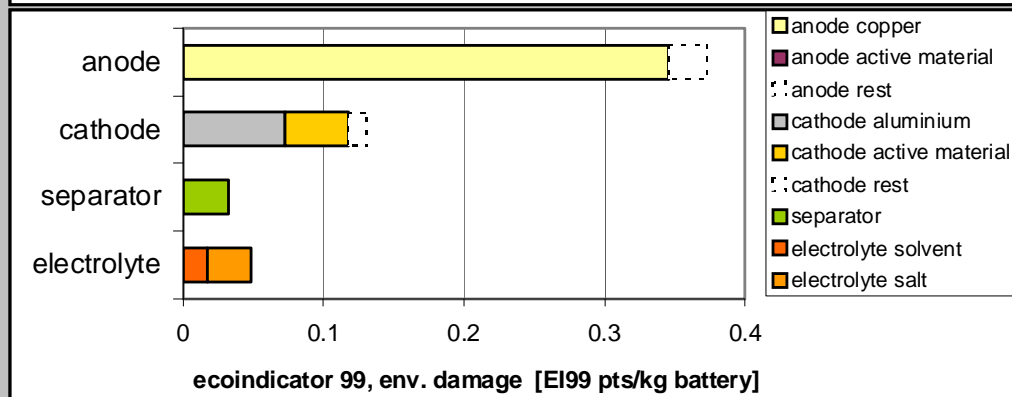
cell composition

- main impact 60% from anode
- 21% from cathode
- rest: cell plant, pouch (plastic), production energy, transport



cell details

- dominant impacts from metals: Cu and Al
- similar impacts from LiMn₂O₄, separator and electrolyte salt LiPF₆



conclusion:
optimisation of metals use



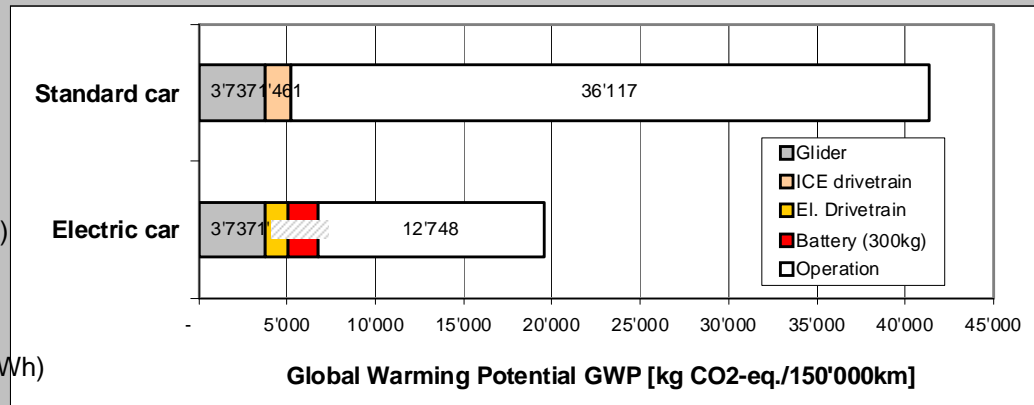
LCA: Entire lifecycle Comparison ICE / BEV vehicle

Global warming potential (kg CO₂-eq) over entire lifecycle (production + 150'000km operation)

ICE, gasoline EURO4
avg. europ. car (golf-class)

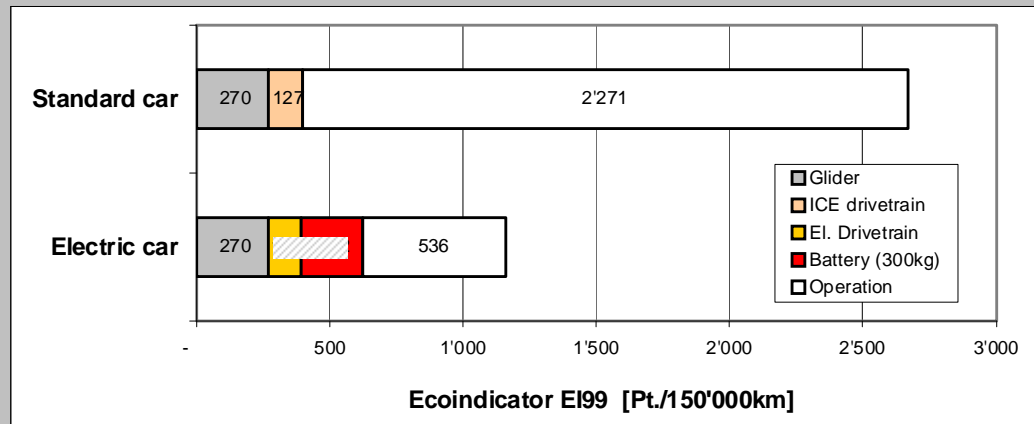
BEV, EU-electricity
(UCTE-mix, 593 gCO₂eq/kWh)
300kg battery

(CH-mix, 134 gCO₂eq/kWh)
(EU-coal-mix, 1095 gCO₂eq/kWh)



minus ~53%

Total environmental impact (EI99 pts) over entire lifecycle (production + 150'000km operation)



minus ~57%

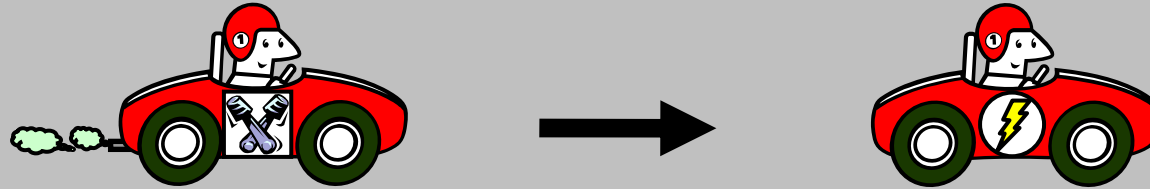
- Due to the higher efficiency in the operation there is a significant advantage for the BEV over the whole lifecycle, even if operated with electricity including a relatively high fossil energy fraction

Conclusion:

your feedback is welcome!

marcel.gauch@empa.ch

- the transition from ICE cars to BEVs looks favorable from an environmental perspective although the impact from BEV production is significantly higher than from ICE car production



- very efficient ICE cars might be competitive with BEVs operated with electricity with very high fossil footprint (pure coal power plants)
- operated with low fossil energy containing electricity, BEVs perform better than the best possible ICE car

